School Based Board Question Paper 2015

SUMMATIVE ASSESSMENT - II (2014-15) SCIENCE Class - X

BFRUVW3

Maximum Marks: 90

Time allowed: 3 hours

General Instructions:

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(1)	The question paper comprises of two Sections, A and B. You are to attempt both the sections.	
(ii)	All questions are compulsory.	
(iii)	There is no choice in any of the questions.	
(iv)	All questions of Section-A and all questions of Section-B are to be attempted separately.	
(v)	Question numbers 1 to 3 in Section-A are one mark questions. These are to be answered in one word or in one sentence.	
(vi)	Question numbers 4 to 6 in Section-A are two marks questions. These are to be answered in about 30 words each.	
(vii)	Question numbers 7 to 18 in Section-A are three marks questions. These are to be answered in about 50 words each.	
(viii)	Question numbers 19 to 24 in Section-A are five marks questions. These are to be answered in about 70 words each.	
(ix)	Question numbers 25 to 33 in Section-B are multiple choice questions based on practical skills. Each question is a one mark question. You are to select one most appropriate response out of the four provided to you.	
(x)	Question numbers 34 to 36 in Section-B are questions based on practical skills. Each question	
	is a two mark question. SECTION-A	
The fo	ormula of citric acid is shown below:	1
THE IC	COOH	7
H ₃ C-	СН ₂ -C-СООН SUNIL TUTORIAL CH ₂ COOH	
	CH ₂	
	the name of —COOH functional group in citric acid.	
Name	the part of the embryo which gives rise to future shoot.	1
How	does concentration of a pesticide change once it enters a food chain.	1
	liary muscle of a normal eye are in their	2
(a) (b)	most relaxed 40 inc. most contracted state. Mention in brief how focal length and power of eye lens will	
	e in two cases. Give reason for the same.	
List fo	our products produced by burning of coal. (62, 21,52, 420.	2
Thoug	th the kulhads are made up of clay which is an eco-friendly substance but their use has	2
been	discontinued in the trains now-a-days. Explain those reasons which lead to	
	atinuance of this practice.	
Carbo this sta	n can neither form C^{4+} cations nor C^{4-} anions but form covalent compounds. Justify attement. Also give reasons why covalent compounds are poor conductor of electricity.	3
An or	ganic compound with molecular formula C ₃ H ₈ O reacts with sodium metal to produce	3
hydro; equati	gen gas. Deduce the possible structures of the compound. Write the balanced chemical on of the reaction.	
	n why the size of the atom increases down the group of periodic	3

16	Name two elements which show the same type of chemical property as calcium. Give reason for the selection of elements along with any one chemical property.	3
11		3
	Which type of organs are shown in the figure above?	
12	(b) Which type of origin and structure do these organs have?	-
123	It was observed that in a family a woman has only daughters. Analyse on the basis of genetics and give an explaination.	3
13	Why does Planaria reproduce by regeneration? Explain the process.	3
14	Write the exocrine and endocrine function of ovary. Where the fertilisation takes place in the human female reproductive system How does the uterus prepare itself to receive and nurture the growing embryo?	3
45	How does sexual reproduction leads to similarities as well as variations?	3
		3
16	With the help of ray diagram explain the formation of rainbow.	3
17	(a) A wooden stick is partly immersed obliquely first in water and then in turpentine oil. In which liquid does the stick appear to be bent more? Give reason in support of your answer. Given that the absolute refractive index of water is 1.33 and that of turpentine	3
	is 1.4.	
	(b) Define absolute refractive index of a medium in terms of velocity of light. How is	
10	relative refractive index different from it?	2
18	Eco-club of your school is organising a debate on the topic 'Nature's Fury Unleashed by	5
	Human's Greed' (Uttrakhand Disaster). (a) List three arguments that you will use to convince the audience that humans are	
	responsible for this disaster.	
	(b) List any three values that will be inculcated with this debate.	1
19	H, Li, Na and K are the elements of same group of Modern Periodic Table.	5
	(a) Arrange them in increasing order of their atomic size	_
	(b) How many valence electrons would each have?	
	(c) How many shells are present in each?	
	(d) Which amongst them is most electropositive?	
20	(a) Why do we say that homozygous plants produce pure progeny?	5
	(b) Define heterozygous.	
	(c) Explain how the process of speciation takes place. FR @2015 for SA2	
2/1	Define the following processes :	5
	(a) Fertilization JSUNIL TUTORIAL	
	(b) Vegetative propagation	
	(c) Menstruation www.jsuniltutorial.weebly.com/	
	(d) Regeneration	
. ,	(e) Binary fission	28
3/2	(a) Explain the term power of accommodation.	5
	(b) A person finds that visibility of things is decreasing for him. Doctor diagnoses that he was suffering from cataract. What happens in this defect of vision and how can it be corrected?	
	(c) List three eye defects.	
	A construction of the cons	

23	(a)	Define power of lens and 1 dioptre.	5
	(b)	What is meant by converging and diverging lens? Explain with the help of figure.	
	(c)	An object 10 cm tall is placed on principal axis of a convex lens. Its 10 cm tall image is	
	(-)	formed on the screen placed 20 cm away from from the lens. Find the focal length.	
24	(a)	Write relation between u, v, f for lens and for mirrors where u, v, f are object distance,	5
<i>J</i> .	()	image distance and focal length respectively.	
	(b)	The magnification produced by a concave mirror is $m = +4$. Write the information	
	(0)	about the image given by this statement.	
	(c)	Draw a ray diagram for the following and show the formation of the images in case of	
	(0)	concave mirror when the object is placed:	
		(i) Between the pole and focus point	
	(ii)	at the centre of curvature	
	(11)		
		SECTION - B	
25	Soap	is chemically:	1
	(a)	mixture of aluminium salts of higher fatty acids.	
	(b) ·	mixture of sodium salts of higher fatty acids.	
	(c)	mixture of calcium salts of higher fatty acids.	
	(d)	mixture of magnesium salts of higher fatty acids.	
26	11 11 11 11 11 11	saponification reaction, fats and oils are treated with:	1
	(a)	A strong acid (b) A weak base	
	(c)	A weak acid (d) A strong base	
27		water that lathers well with soaps is :	1
	(a)	Hard water (b) Soft water	_
	(c)	Sea water (d) Bromine water	
28	The second secon	bject is placed in front of a concave mirror at infinity. What is the nature of the image?	1
	(a)	Virtual and inverted (b) Virtual and erect	-
	(c)	Real and erect (d) Real and inverted	
29		l length of the concave mirror depends on its:	1
	(a)	Size (b) Aperture	_
	(c)	Radius of curvature (d) Material used	
30	- 1	eperiment of tracing the path of a ray of light through a glass slab was set up in the laboratory and	1
50	rav di	agram was drawn as shown OLIESTION DADED GOOLE for SAO	1
	P	agram was drawn as shown. QUESTION PAPER @2015 for SA2	
		JSUNIL TUTORIAL	
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		www.jsuniltutorial.weebly.com/	
	1		
	: 1	N_3	
	. 1	$N_2 \downarrow_R$	
		`S	
		N_4	
	The re	efracted ray in this diagram is :	
	(a)		
	(c)	PQ (b) $QR \cdot N_1N_2$ (d) RS	
31			
	of ray	h of the following figures has been labelled correctly in an experiment to trace the path is of light through a glass prism	1

angle of incidence 'Normal Normal angle of refraction angle of incident (a) incident ray fig.1 angle of refraction Normal Normal (b) angle of emergent angle of incidence incident ray fig.2 angle of deviation Normal - Normal angle of incident angle of emergence (c) incident ray fig.3 angle of emergent Normal - Normal angle of incident angle of prism (d)

