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Code No. RSPL/1

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Candidates must write the Code on the title page of the answer-book.

- · Please check that this question paper contains 8 printed pages.
- Code number given on the right hand side of the question paper should be written on the title page of the answer-book by the candidate.
- Please check that this question paper contains 42 questions.
- Please write down the Serial Number of the question before attempting it.
- 15 minutes time has been allotted to read this question paper.

SUMMATIVE ASSESSMENT-II

SCIENCE

Time allowed: 3 hours]

I Maximum Marks: 90

General Instructions:

- The question paper comprises of two sections, A and B. You are to attempt both the sections.
- (ii) All questions are compulsory.
- (iii) All questions of Section-A and all questions of Section-B are to be attempted separately.
- (iv) Question numbers 1 to 3 in Section-A are one mark questions. These are to be answered in one word or in one sentence.
- (v) Question numbers 4 to 7 in Section-A are two marks questions. These are to be answered in about 30 words each.
- (vi) Question numbers 8 to 19 in Section-A are three marks questions. These are to be answered in about 50 words each.
- (vii) Question numbers 20 to 24 in Section-A are five marks questions. These are to be answered in about 70 words each.
- (viii) Question numbers 25 to 42 in Section-B are multiple choice questions based on practical skills. Each question is one mark question. You are to select one most appropriate response out of the four provided to you.

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SECTION-A

1.	Detailed study of forelimbs of mammals, birds, reptiles and amphibians shows that the bones of all these have same structure and points towards a common origin.
	(i) What conclusion can be drawn from this?
	(ii) What is the term given to such forelimbs?
2.	Draw the electron dot structure of ethyne and also draw its structural formula.
3.	What is the information source for making proteins in the cell? State two basic events in reproduction.
4.	(i) Why does the sun appear reddish at sunrise?
	(ii) What is Tyndall effect?
5.	Name the element(s) of period 3 of modern periodic table which
	(i) are classified as metals.
	(ii) have largest atomic size.
	(iii) form(s) negative ions.
	(iv) exist(s) as gases,
6.	List any four advantages of wefer harvesting

7. The following table shows the absolute refractive index of some material media:

S.No.	Material medium	Refractive index
A	Air	1.0003
В	Alcohol	1.36
C	Benzene	1.50
D	Rock salt	1.54
Ē	Ruby	1.71

Using the above table, answer the following questions:

- (i) What change takes place in the direction of a ray of light when it passes from benzene to alcohol?
- (ii) In which medium does the light travel the slowest? Justify your answers in each case.

8. (i) State laws of refraction of light.

- (ii) The speed of light in a transparent medium is 0.7 times that of its speed in vacuum, Calculate the absolute refractive index of the medium.
- 9. (i) Why do covalent compounds have low melting and boiling points?

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(ii) Write IUPAC name of the following:

(iii) Name the gas evolved when methanol reacts with sodium metal. Write the chemical reaction.

10. How is the ozone formed in higher levels of atmosphere? Which chemicals are responsible for the depletion of ozone layer?

- 11. (i) Define magnification produced by a lens. What does the positive sign of magnification signify?
 - (ii) A convex lens of focal length 40 cm and a concave lens of focal length 50 cm are placed in contact with each other. Calculate the (a) power, and (b) focal length of this lens combination.

12. Table given below is part of periodic table :

1	2	13	14	15	16	17	18
H							He
Li	Be	В	C	N	0	F	Ne
Na	Mg	Al	Si	P	S	Cl	Ar
K	Ca	Ga	Ge	As	Se	Br	Kr

- (i) Which group elements are reactive metals and why?
- (ii) Which of them has largest atomic size and why?
- (iii) Which of the element is most reactive in group 17 and why?

What are sexually transmitted diseases? Give example of two bacterial and two viral sexually transmitted diseases.

- 14. Srishty, a student of class X, met her friend who was in class IX on her way to home after school. Srishty noticed that her friend was finding it difficult to read the bus numbers and helped her to get into the bus. She enquired and found that her friend had a defect in her eyesight and had difficulty in viewing far away objects and also, she was not in a position to afford a check-up with an eye specialist. She immediately took her to a nearby ophthalmologist. The doctor advised her friend to wear spectacles and gave her the power of spectacles to be worn. She paid the doctor's fee from her pocket money/savings and also purchased medicines prescribed by the doctor for her friend.
 - (i) What values do you see in Srishty?
 - (ii) Explain the eye defect by which Srishty's friend was suffered and also, give the rectification to be adopted by ophthalmologist with a ray diagram.

15. The atomic number of an element is 15.

- (i) Identify the element.
- (ii) · Write its electronic configuration.

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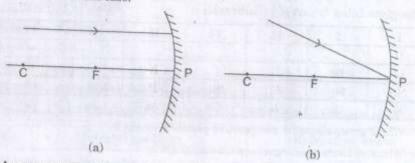
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- (iii) Write its valence electrons and valency.
- (iv) Identify the nature of oxide formed by it.
- (v) Give the formula of its chloride.
- (vi) Identify the group and period to which it belongs.
- 16. Giving an example, differentiate between acquired and inherited traits.
- 17 State and describe in brief any three main factors responsible for the rise of a new species.
- 18. (i) What are the consequences of depletion of forests?
 - (ii) What is the advantage of a disposable paper cup over a plastic cup?
 - (iii) Why is reuse better than recycling in conserving our environment?
- 19. Why is vegetative propogation practised for raising some types of plants? List any two advantages and disadvantages of vegetative propogation.
- 20. (i) Complete the following ray diagrams. Also mark the angle of incidence and angle of reflection in each case.



- (ii) A concave mirror forms an inverted image of size double than that of object. The focal length of the mirror is 15 cm. Find the position of the image and object.
- 21. A compound 'A' (molecular formula C3H6O2) reacts with NaHCO3 to form a compound 'B' and evolves a gas which extinguishes fire, Compound 'A' on treatment with alcohol 'C' in presence of conc. H_2SO_4 forms a compound 'D' (molecular formula $C_5H_{10}O_2$). On addition of NaOH to 'A', it also gives 'E' and water. 'D' on treatment with dilute NaOH solution gives back 'A' and 'C'. Identify A, B, C, D and E and write down the reactions involved.
- 22. Draw a neat diagram of human female reproductive system and label the following :
 - (i) Part which produces female germ cell.
 - (ii) Part where fertilisation takes place.
 - (iii) Part where zygote is implanted.

What changes take place in the lining of uterus

- (i) before implantation,
- (ii) if no fertilisation occurs?

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23.	(62	What is meant by least distance	of distin	nct vision? Mention its	value.	
	(ii)	In which of the following two case	ses, the	focal length of the eye	lens will be more -	
		(a) when ciliary muscles of a no	rmal ey	e is in most relaxed st	ate, or	
		(b) when ciliary muscles of a no	rmal ey	e is in most contracted	I state.	
		Explain with reason.				
	(iH)	Newton placed two triangular p He then let white light pass thro reason for such observation and	ugh ther	n. What was his observ	ration on screen? State	5
24.	Dri	nw a diagram of a pistil showing a	ermina	tion of pollen and labe	the following parts:	
	(1)	Part that transfers male gamete	to ovul	0.		
	(ii)	Part which fuses with female ga	mete.			
	(iii	Part which develops into a fruit.				
	Lis	t the various changes that take p	lace in fl	lower after fertilisation).	5
			SECTI	ON-B		
25,		nanoic acid is added to sodium blo ming splinter. The following four of			evolved is tested with a	
	I.	The gas burns with a pop sound	and the	flame gets extinguish	ed.	
	II.	The gas does not burn but the sp	olinter b	urns with a pop sound		
	III.	The flame extinguishes and the	gas does	not burn.		
	IV.	The gas burns with a blue flame	and the	splinter burns bright	ly.	
	The	e correct observation is reported i	n			1
	(n)	I (b) II		(c) III	(d) IV	
26.	On	adding acetic acid to sodium met	al,			1
	(a)	greenish yellow gas with punger	nt smell	evolves.		
	(b)	colourless and odourless gas evo	lves whi	ich burns with 'pop' so	und.	
	(c)	yellow coloured odourless gas ev	olves.			
	(d)	colourless gas with pungent sme	ell evolve	99.		
27.	Soc	ip works well in				1
	(a)	hard water	(h)	soft water		
	(e)	acidified water	(d)	very cold water.		
28.	Cor	mmon salt is added in preparati	on of so	ap so as to		1
	(n)	favour precipitation of soap	(b)	act as filler		
	(c).	prevent precipitation of soap	(d)	make it salty and inc	rease its solubility.	

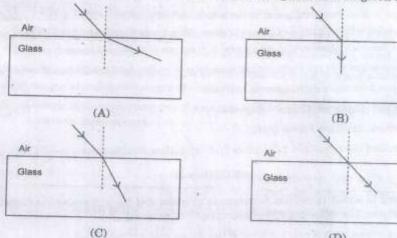
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- 29. Soaps are cleansing agents and are
 - (a) sodium salts of fatty acids
- (b) calcium salts of fatty acids
- (c) potassium salts of mineral acids
- (d) magnesium salts of fatty acids.
- 30. The reaction involved in preparation of soap is called
 - (a) esterification

(b) saponification

(c) hydrogenation

- (d) hydration.
- 31. Four students traced the path of a ray of light from glass to air as given below.

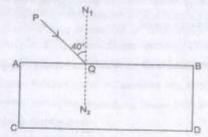


The correct path of refracted ray is shown in

- (b) (B)
- (c) (C)

(D)

- 32. While performing an experiment to trace the path of a ray of light passing through a rectangular glass slab, a ray PQ is made to incident at an angle of 40° by a student on the side AB of glass slab as shown below.



The angle of refraction observed by him would be

(a) more than 40°

(b) less than 40"

(c) equal to 40"

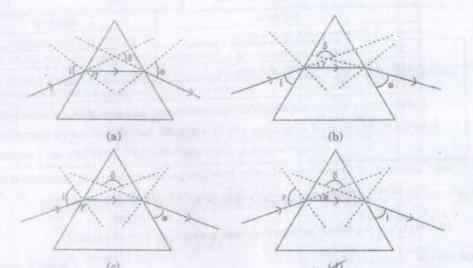
(d) equal to 0°



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38. Which of the following figures has been labelled correctly in an experiment to trace the path of ray of light through a glass prism?



- 34. To determine the focal length of a concave mirror, four students P, Q, R and S obtained the image of an electric pole on a wall. They measured the distance as given below between :
 - P electric pole and the wall only
 - Q electric pole and the mirror only
 - R mirror and the wall only
 - S electric pole and the wall and also between the electric pole and the mirror. The correct focal length will be directly obtained by

(a) P

(c) R

- (d) S
- 35. In the laboratory, a student was provided with a convex lens whose focal length was somewhere between 15 cm to 20 cm. While focusing a distant object to find out its focal length, he should adjust the position of screen between

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- (a) 15 cm and 20 cm marks on scale from lens
- (b) 20 cm and 30 cm marks on scale from lens
- (c) 25 cm and 30 cm marks on scale from lens
- (d) 30 cm and 40 cm marks on scale from lens.
- 36. A virtual erect and magnified image of an object is formed by a convex lens. The position of the object is

- (a) between F and 2F
- (b) between 2F and infinity

(c) at the focus

(d) between F and optical centre.

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37. The given slides A and B were identified by four students I, II, III and IV as stated below:







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(B)

	Slide A	Slide B
1	Binary fission in Amoeba	Daughter cells of Amoeba
II	Budding in yeast	Buds of yeast
Ш	Binary fission in Amoeba	Buds of yeast
IV	Budding in yeast	Daughter cells of Amoeba

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	Oft	he above mentioned identifications of	f slie	des A and B, which one is correct?	-
	(a)	I		П	
	(e)	III	(d)	IV	
38.	The	mode of asexual reproduction in wh	ich p	arental identity is maintained is known as	
	(a)	binary fission		multiple fission	
	(c)	budding	(d)	all of these	
39.	Whi	ch of the following organs is analogo			
		Flipper of whale		Foreleg of horse	
	(e)	Wings of housefly		Front leg of frog.	
40	Whi	ch of the following on the			

- 40. Which of the following are the examples of homologous organs?
 - (a) Venus flytrap and spines of cactus
 - (b) Tendrils of cucurbits and venus flytrap
 - (c) Spines of barberry and fruit of lemon
 - (d) Venus flytrap and pitcher plant
- 41. The part of embryonal axis between plumule and point of attachment of cotyledons is called
 - (a) epicotyl

(b) hypocotyl

(c) radicle

- (d) hiltum.
- 42. In which of the following options, the part of the seed is incorrectly matched with its characteristics?
 - (a) Hilum point of attachment of seed stalk
 - (b) Radicle develops into future root
 - (c) Micropyle absorbs water during germination
 - (d) Seed coat stores food for developing embryo