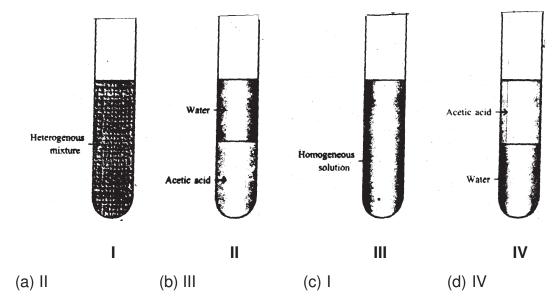
Experiment - 15

Properties of Acetic Acid

Aim: To study the following properties of acetic acid (ethanoic acid, CH₃COOH)

- (i) odour
- (ii) solubility in water
- (iii) effect on litmus paper
- (iv) reaction with sodium bicarbonate
- Q. 1 Which one of the following would be observed when 10ml of acetic acid is added to10ml of water?



- Q. 2 Which one of the formula of ethanoic acid (acetic acid)
 - (a) NaHCO₃

(b) Na₂CO₃

(c) CH₃COOH

- (d) CH₃CH₂OH
- Q. 3 Which one of the following has vinegar like odour?
 - (a) Sodium bicarbonate

(b) Ethanol

(c) Acetic acid

- (d) sodium carbonate
- Q. 4 On adding blue litmus solution to acetic acid the colour changes to
 - (a) Red

(b) Blue

(c) Orange

(d) No change in colour

	(c) Acidic	;		(d) Amphoteric						
Q. 6	Which is the correct statement for the solution when 2ml of acetic acid is mixed with water?									
	(a) Trans	parent		(b) Precipitate						
	(c) Turbic	d		(d) Separate lay	er of acid					
Q. 7	(a) should(b) should(c) may b	sting the effect of vapours of acetic acid on litmus paper, the litmus paper :- ould be moist ould be dry ay be moist or dry it does not matter es not give any test								
Q. 8	of a gas (a) sugar (b) comm (c) Sodiu	nl of acetic acid solution is taken in a test tube. On adding a pinch of 'X' effervescence a gas 'y' is produced, which turns lime water milky. Name 'X' and 'y'.) sugar and CO ₂ gas respectively) common salt and H ₂ gas respectively) Sodium bicarbonate and CO ₂ has respectively) Hydrochloric acid more hydrogen gas								
Q. 9	Yamini studies the properties of acetic acid in laboratory and recorded the observation as given in table. Which one is correct record of the following :- Option Odour Solubility in water									
		Vinegar	Lemon	Partialy soluble	Soluble					
	а	~	×	~	×					
	b	~	×	×	✓					
		×	y	×						

(b) Neutral

Q. 5 What is the nature of acetic acid?

X

(a) Basic

×

Q. 10 Vasu studies the chemical properties of acetic acid in laboratory. He recorded the observations in the table given below. The correct option of result would be in :-

Option	effect on litmus		Reaction with NaHCO ₃		
	Vinegar	Lemon	Partialy soluble	Soluble	
а	✓	×	~	×	
b	✓	×	×	~	
С	×	~	×	~	
d	×	~	~	×	

Q. 11 A student added acetic acid to test tube I, II, III and IV and then introduced a burning candle near the mouth of each test tube. In which case of candle would not be extinguished?

